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| --- |
| **Give correct names for these ionic compounds**  |
| KCl | MgO |
| K2O | AlCl3 |
| CaO | CaS |
| MgCl2 | Al2S3 |
| NaH | BeF2 |
| ZnS | MgI2 |
| Al2O3 | BaBr2 |
| Na3N | CsCl |
| Ca2C | Mg3P2 |
| KI | CaCl2 |

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| **Give correct formulas for these ionic compounds**  |
| calcium fluoride | magnesium phosphide |
| calcium hydride | sodium chloride |
| magnesium fluoride | sodium oxide |
| magnesium bromide | aluminum phosphide |
| sodium nitride | calcium sulfide |
| lithium oxide | potassium oxide |
| barium nitride | sodium fluoride |
| lithium chloride | lithium sulfide |
| aluminum nitride | sodium hydride |
| aluminum fluoride | magnesium nitride |